Quantum Mechanics and Atomic Physics Lecture 4: Schodinger's Equation: Part II

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Announcement

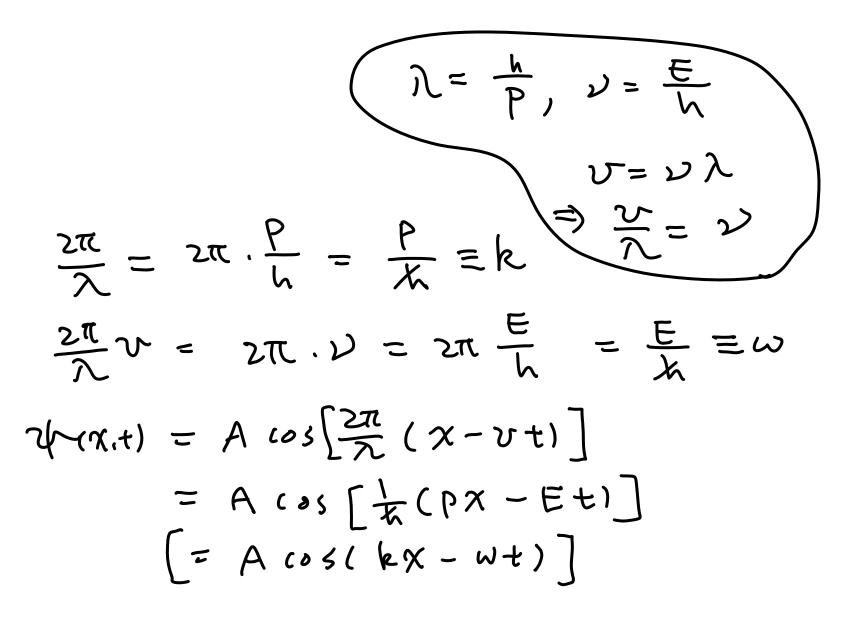
Second HW will be due on Monday Sept 19!

The text book is on reserve in SERC reading room

Let's recap

- Last time we "derived" time-independent Schroedinger's equation!
 - $-\frac{k^{2}}{2m}\frac{\partial^{2}}{\partial x^{2}}\psi + V.\psi = E\psi$
- We derived this using $\lambda = \frac{h}{P}$ and a prototype wave function $\psi(x,t) = A \cos(\frac{2\pi}{2}(x-vt))$
- Today we will derive time-dependent Schroedinger equation
- $-\frac{k^2}{2m}\frac{\partial^2}{\partial x^2}\psi(x,t) + V(x,t)\cdot\psi(x,t) = \frac{\partial^2}{\partial t}\psi(x,t)$
- And discuss the probability interpretation of the wavefunctions

Remember



Strategy for time-dependent Schroedinger Equation

From
$$\frac{\beta^2}{2m} + V = E$$
 and
 $\psi(x,t) = A\cos\left(\frac{1}{4k}(px - Et)\right),$
Differentiation by $\psi(x^2)$ allowed nonlocation

- Differentiating by "x" allowed replacement of "p" and led to time-independent Schroedinger equation.
- So differentiating by "t" will lead to replacement of E and to the time-dependent Schroedinger equation.

Strategy

So our goal is to replace "E" by time-derivative 111 $\Gamma \frac{P}{2m} + V = E$

Let's try to include time-dependence

Problems with this result

- Amplitude appears in what we presume to be a general physical law
 - A should be dictated by the "boundary conditions" of the problem (more on this next time)
- There is a sign ambiguity
- There is a square root in the denominator!
 - This differential equation is non-linear
 - A sum of independent solutions would not itself be a solution

Plane wave representation for a free particle

- Schroedinger had the idea to modify the wave equation
- So the free particle wave equation becomes:

$$\Psi(x,t) = A \cos\left[\frac{1}{h}(px-Et)\right] + iA \sin\left[\frac{1}{h}(px-Et)\right]$$

= $Ae^{i\left[\frac{1}{h}(px-Et)\right]}$

Time-Dependent S.E.

$$\frac{\partial \Psi}{\partial t} = A e^{\frac{i}{\hbar}(px-Et)} \left(-\frac{i}{\hbar}E\right) = -\frac{i}{\hbar}E\Psi$$

$$\frac{\partial \Psi}{\partial x} = A e^{\frac{i}{\hbar}(px-Et)} \left(\frac{i}{\hbar}p\right) = \frac{i}{\hbar}P\Psi$$

$$\frac{\partial \Psi}{\partial x^{2}} = \frac{i}{\hbar}P \frac{\partial \Psi}{\partial x} = \frac{i^{2}}{\hbar^{2}}P^{2}\Psi = -\frac{p^{2}}{\hbar^{2}}\Psi$$

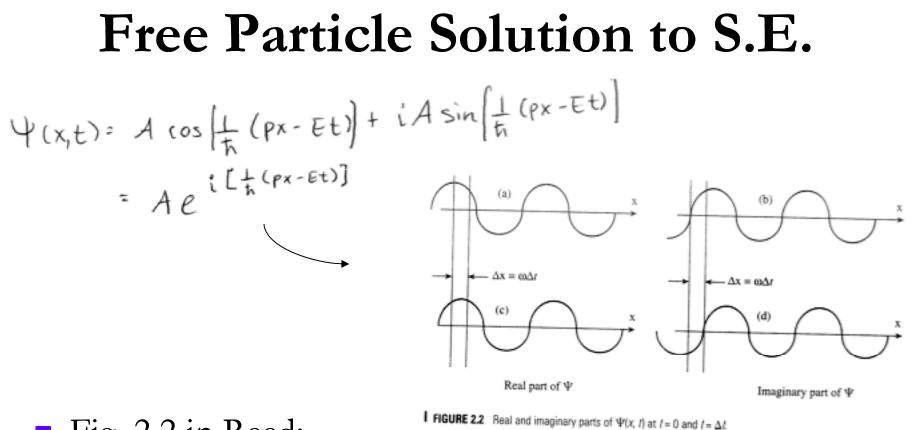
$$Now E = \frac{p^{2}}{2m} + V(x,t)$$

$$\Rightarrow E\Psi = \frac{p^{2}}{2m}\Psi + V\Psi$$

$$-\frac{\hbar}{i} \frac{\partial \Psi}{\partial t} = -\frac{\hbar^{2}}{2m} \frac{\partial^{2}\Psi}{\partial x^{2}} + V\Psi$$

$$wse: \frac{1}{t} = \frac{1}{t} \cdot \frac{i}{t} = \frac{i}{t^{2}} = -i$$

$$\Rightarrow \begin{bmatrix} -\frac{\hbar^{2}}{2m} & \frac{\partial^{2}\Psi}{\partial x^{2}} + V\Psi = i \ln \frac{\partial\Psi}{\partial t} \end{bmatrix}$$
Time Dependent Schrödingers Equation!



- Fig. 2.2 in Reed:
 - (a), (b): real (cosine) and imaginary (sine) parts of Ψ at t=0.
 - (c), (d): at a slightly later time Δt
 - The pattern for Ψ behaves as if it has advanced slightly to the right between t=0 and t = Δ t.
 - Similarly for motion to the left (~ $e^{-i(kx+\omega t)}$).

Born's interpretation

- Question: how are we to interpret a QM wavefunction Ψ?
 - Interpretation in terms of probabilities by Max Born in 1926.

Wavefunction Ψ

- $\Psi(x,t)$ is a general complex number: $\Psi=A + iB$ or $Re^{i\theta}$ where $i=\sqrt{-1}$.
- Probability of finding particle between x and (x+dx) at time t is:

 $P(x,t)dx = \Psi^*(x,t) \Psi(x,t)dx$ (Born's Interpretation) where $\Psi^*(x,t)$ is the complex conjugate of Ψ

 $\Psi^*(x,t) \Psi(x,t)$ is known as the *probability density*

 $P(x,t) = \Psi^*(x,t) \Psi(x,t) = (A-iB)(A+iB) = A^2 - i^2B^2 = A^2 + B^2$ So P(x,t) is real and positive!

Born's interpretation, con't

- Probabilistic interpretation of Ψ does not mean that a particle can be in two or more places simultaneously!
 - Particles do not loose their identity as (essentially) point objects
- In this interpretation, we can only predict the probability of finding the particle in some region of space or of its having momentum within certain limits.
- Ψ itself has no physical reality it's a mathematical construction to keep account of probabilities.
- The "waviness" of matter is a consequence of its probability distribution.

Normalization

We say that a wavefunction is normalized if:

$$\int_{-\infty}^{\infty} P(x,t) dx = 1 \implies \int_{-\infty}^{\infty} \Psi^*(x,t) \Psi(x,t) dx = 1$$

- If one adds up the probabilities of finding a particle over the entire space it could possibly occupy, the total must be unity.
- The probabilities must add up to a "whole particle"
- In other words, the wavefunction Ψ must be square integrable
 - This is one of the **boundary conditions** that must be imposed on solutions to the S.E.!

Example

• A free particle's wave function is represented by

$$\psi(x,t) = A e^{\frac{1}{2}(p_{\pi} - Et)}$$

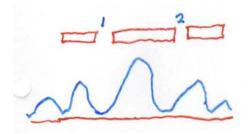
Which one of the following depicts the probability of finding the particle correctly?
 (G)
 (G)

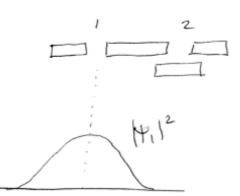
Recall: Linearity

- If $\Psi_1(x,t)$ and $\Psi_2(x,t)$ are two different solutions to the equation for a given potential energy V, then any arbitrary linear combination of these solutions, $\Psi(x,t) = c_1 \Psi_1(x,t) + c_2 \Psi_2(x,t)$, is also a solution.
- It involves the first (linear) power of $\Psi_1(x,t)$ and $\Psi_2(x,t)$
- c_1 and c_2 can have any (arbitrary) complex values
- Linearity ensures we can add together wave functions
 - Constructive and destructive interference
 - Principle of superposition

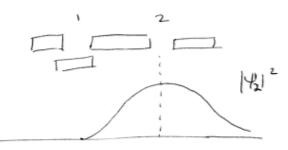
Principle of superposition: Double-slit experiment

Only slit 1 open, we have: \(\Psi_i (x_i t) = R_i (x_i t) e^{i \(\theta_i)}\)
Probability distribution with only slit 1 open: \(\Psi_i = \(\Psi_i^* t)_i = R_i e^{-i \(\theta_i)} R_i e^{i \(\theta_i)} = R_i^2\)





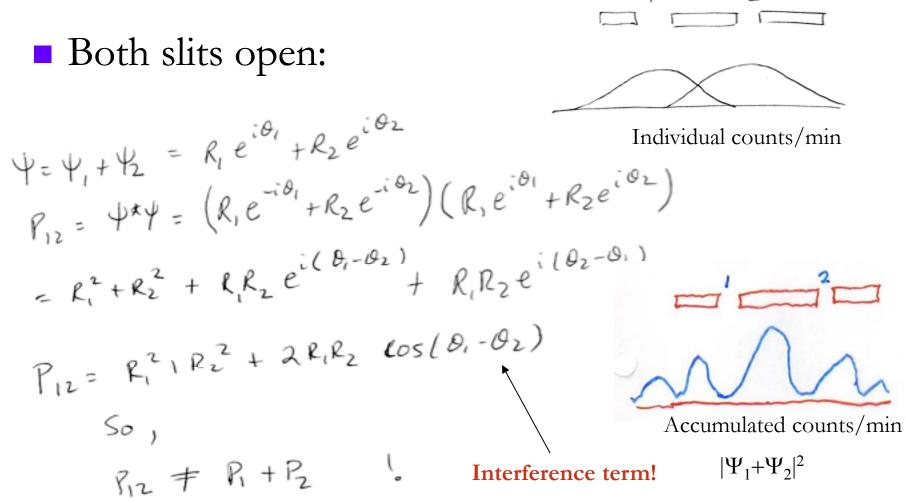
Probability distribution



Probability distribution

If only slit 2 open: $\Psi_2 = R_2 e^{i\theta_2} \implies P_2 = \Psi_2^* \Psi_2 = R_2^2$

Principle of superposition, con't



NOT $|\Psi_1|^2 + |\Psi_2|^2$

Boundary Conditions

- 1. As we already covered, Ψ must be square integrable: $\int_{0}^{\infty} \rho(x,t) dx = I \implies \int_{0}^{\infty} \Psi^{*}(x,t) \Psi(x,t) dx = I$
- 2. The wavefunction Ψ must be a continuous function!
 - This means forcing two solutions at the boundary to agree: $\Psi^{<}(\text{boundary}) = \Psi^{>}(\text{boundary})$

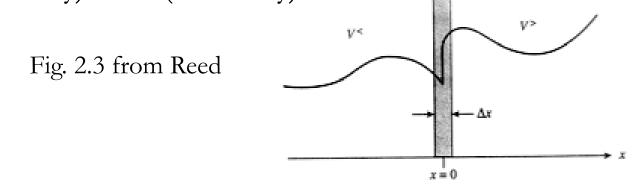


FIGURE 2.3 Potential function with a discontinuity at x = 0.

- 3. If V(x) is continuous or finitely discontinuous across a boundary, then the first derivative of Ψ , $d\Psi/dx$, must be made continuous across the boundary. But if V(x) is infinitely discontinuous across the boundary, then $d\Psi/dx$ can be discontinuous across the boundary.
- These will become clear once we start doing some examples.

Error in the book

In Eq. 2.4.7, $\nabla^{7}(o) - \nabla^{7}(o)$ $\Rightarrow \nabla^{7}(o) + \nabla^{7}(o)$

Change of Potential by a constant amount

- Suppose we have a particle with potential V(x), with a wavefunction solution Ψ and energy state E.
- If the potential changes to V'(x) = V(x) +V₀, where V₀ is constant, what will be new Ψ' and E'?

$$-\frac{h^2}{am}\frac{d^2\Psi}{dx^2} + V(x)\Psi = E\Psi$$
(1)

New:

$$-\frac{h^{2}}{2m}\frac{d^{2}}{dx^{2}}\Psi' + [V(x)+V_{0}]\Psi' = E'\Psi'$$

$$\Rightarrow -\frac{h^{2}}{2m}\frac{d^{2}\Psi'}{dx^{2}} + V(x)\Psi' = (E'-V_{0})\Psi' (2)$$

... continued

- (1) and (2) will be identical if $E=E'+V_0$.
- So the new wavefunction will be identical $\Psi' = \Psi$. But the new energy state will be $E=E'+V_0$.
- All that happens is that the "zero" of the potential has shifted.
- This is true for all wavefunction solutions, Ψ_n (or *eignefunctions*) and E_n (or *energy eignevalues*). More on eigenfunctions and eigenvalues next time.

Example: "particle-in-a-box"

- Consider a particle of mass m which can move freely along the x axis anywhere from x=-a/2 to x=+a/2, but is strictly prohibited from being found outside this region. The particle bounces back and forth between the walls at x=±a/2 of a (1-dim) box. Assume the walls to be completely impenetrable, no matter how energetic the particle is (this is an idealization!).
- The wave function of the particle is:

$$\Psi(\mathbf{x},t) = A\cos\frac{\pi x}{a} e^{-iEt/\hbar} -\frac{9}{3}cxcd$$

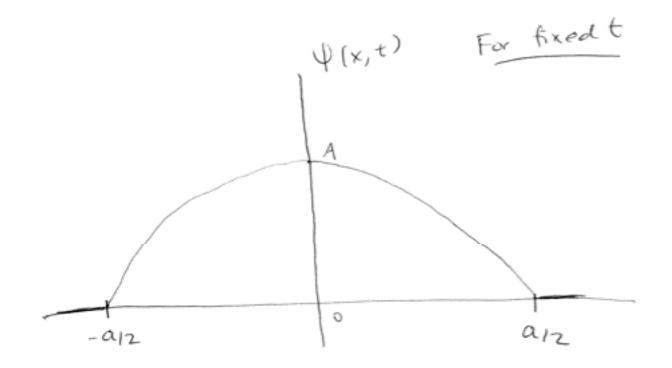
Example, con't

• Verify that it is a solution to the S.E. in the region $-a/2 \le x \le +a/2$ and determine the value of the lowest energy state.

Since there are no torces acting on particle, V= constant in the region. : We can take V=0 in the region. So. $-\frac{h^2}{2m}\frac{\partial^2\Psi}{\partial x^2} = i\frac{h}{\partial t}\frac{\partial \Psi}{\partial t} - \frac{a}{2}\langle x \langle t^a \rangle_2$ -iEt/t $\frac{\partial Y}{\partial x^2} = -\left(\frac{\pi}{a}\right)^2 A \cos \frac{\pi x}{a} e^{-iEt/\hbar} = -\left(\frac{\pi}{a}\right)^2 \Psi$ $\frac{\partial \Psi}{\partial t} = -\frac{iE}{t} A \cos \frac{\pi e}{Q} e^{-iEt/t} = -\frac{iE}{t} \Psi$ $\Rightarrow + \frac{h^2}{2} \left(\frac{\pi}{a} \right)^2 \Psi = -i \frac{h}{b} \frac{i E \Psi}{b} \Psi$ $\frac{1}{am}\left(\frac{k\pi}{a}\right)^{2}\Psi = E\Psi \Rightarrow \left(E = \frac{\pi^{2}h^{2}}{2ma^{2}}\right)$

Example, con't

Plot the space dependence of the wave function (for fixed time).



Summary/Announcements

- We derived time-dependent Schroedinger equation
- Born's interpretation of probability density for wavefunctions
- Boundary conditions
- Next time:
 - Solutions to Schrodinger's Equation in 1 dimension

Next homework due on Monday Sept 19!