

Homework 3 Solutions

Physics 406.

1. O chap. 3 Q 3.

In the Einstein model the atoms are treated as independent simple harmonic oscillators (SHOs), with a single frequency ω_E .

By contrast in the Debye approach the atoms interact to produce collective lattice motions (e.g. sound waves) but there is assumed to be no interaction between these waves.

As a result a single wave does not decay or transform with time, and this model does not include thermal expansion.

9. O chap. 3 Q. 14.

The phonon mean-free path is primarily determined by

(i) phonon-phonon scattering

(ii) scattering of phonons by defects
in the solid

(iii) scattering of the phonons by
sample boundaries

If we assume the relation

$$K = \frac{1}{3} c_v n l$$

the phonon mean-free path can be determined by

(a) direct comparison of measurements of $\kappa(T)$ and $c_v(T)$

(b) ultrasound attenuation

(measured attenuation of "phonon beam")

(c) Raman and Brillouin (light) scattering

↑ ↑
optical acoustic
phonon phonon

Rayleigh line - static impurities

Stokes lines - Phmm scattering

3.

8. Chap 3 p. 6.

a) $E = 3RT = 3R\Theta_D$

$$= (3)(8.31 \text{ J K}^{-1} \text{ mol}^{-1})(340 \text{ K})$$

$$= (3)(8.31)(340) \text{ J mol}^{-1}$$

$$\boxed{1 \text{ mol of Cu} \Rightarrow 8.48 \times 10^3 \text{ J}}$$

For comparison to the value that we'll get from the graph in Fig 3.13 (see b), we want this energy in calories \Rightarrow

then

$$E_{\text{class}} = 3RT = 3(2 \text{ cal K}^{-1} \text{ mol}^{-1}) (340) \\ = 2040 \text{ calories mol}^{-1}$$

↓

1 mole of Cu $\Rightarrow E_{\text{class}} = 2040 \text{ calories}$

b)

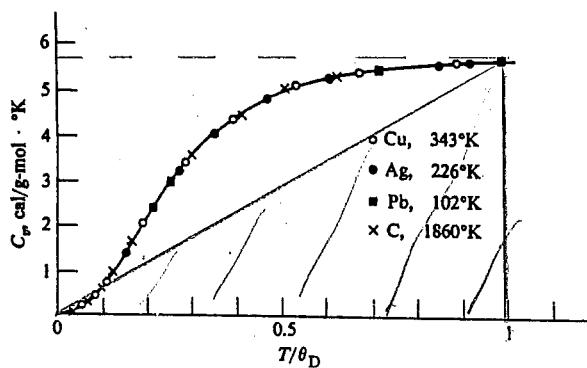


Fig. 3.13 Specific heats versus reduced temperature for four substances. Numbers refer to Debye temperatures. Note the high Debye temperature for diamond.

$$E_{\text{Debye}} \sim \int_0^{\theta_D} dT C_V$$

$$\sim \frac{1}{2} (5.75) (3.40) \sim 977 \text{ calories}$$

Therefore

$$E_{\text{Debye}} < E_{\text{classical.}}$$

c) $E_{\text{Debye}} \sim \langle T \rangle + \langle N \rangle \sim 2 \langle v \rangle$



$$E_{\text{Debye}} \sim 2 \left(\frac{1}{2} m \omega_D^2 \langle x^2 \rangle \right)$$



$$\langle x^2 \rangle \sim \frac{E_{\text{Debye}}}{m \omega_D^2}$$



$$\sqrt{\langle x^2 \rangle} \sim \frac{1}{\omega_D} \sqrt{\frac{E_{\text{Debye}}}{m}}$$

$$[L] \stackrel{?}{=} [T^2] \sqrt{\frac{[M][L]^2 / [T]^2}{[M]}}$$

$$\stackrel{?}{=} [T^2] \frac{[L]}{[T]} \quad \checkmark$$

$$[L] = [L] \quad \checkmark$$

$$E_0 \sim (340) k_B \sim (340)(1.4 \times 10^{-23})$$

$$m \sim \pi m_p \sim (29)(1.6 \times 10^{-27}).$$

$$\omega_0 \sim \frac{k_B T_0}{\hbar} \sim \frac{(1.4) \times 10^{-3}}{1.05 \times 10^{-34}} (340)$$

$$\sim 4.5 \times 10^{13}$$

$$\sqrt{\langle x^2 \rangle} \sim \frac{1}{4.5 \times 10^{13}}$$

$$\sim \frac{10^2}{4.5 \times 10^{13}}$$

$$\sqrt{\frac{(340)(1.4 \times 10^{-23})}{(29)(1.6 \times 10^{-27})}}$$

$$\sqrt{\frac{(340)(1.4)}{(29)(1.6)}}$$

$$\sqrt{\langle x^2 \rangle} \propto 7 \times 10^{-10} \text{ m} = 7 \text{ \AA}$$

This is to be compared w/ the lattice spacing of copper, which is 2\AA . Therefore at $T \sim 0$, the maximum displacement of the atoms is about 3.5 lattice spacings

4.

In the Einstein model

$$U(T) = \sum_n \left(n + \frac{1}{2}\right) \hbar \omega_E$$

$$T=0 \Rightarrow U = \frac{1}{2} \hbar \omega_E \quad \text{since } n=0$$

For a harmonic oscillator $\langle T \rangle = \langle V \rangle$



$$\langle E \rangle = \langle T \rangle + \langle V \rangle = 2 \langle V \rangle$$

$$= 2 \left(\frac{1}{2} m \omega_E^2 \langle x^2 \rangle \right) = \frac{1}{2} \hbar \omega_E$$



$$\langle x^2 \rangle = \frac{k}{2m\omega_E}$$

Typically $\omega_E \sim 10^{13} \text{ s}^{-1}$

Use $m_p \sim 10^9 \text{ eV/c}^2$



$$\sqrt{\langle x^2 \rangle} = \frac{(\hbar c) c}{4\pi (mc^2)(\omega_E)}$$



$$\sqrt{\langle x^2 \rangle} = \frac{(hc) c}{4\pi (mc^2)(\omega_E)} = \frac{(1240.0 \text{ eV}\cdot\text{\AA}) (3 \times 10^{18} \text{ \AA/s})}{(4\pi) (10^9 \text{ eV}) (10^{13})}$$

$$\sim \frac{(1.24)(3)}{4\pi} \frac{10^4 / 10^{18}}{10^9 / 10^{13}}$$

$$\sim .3 \text{ \AA}$$

$\sqrt{\langle x^2 \rangle} \sim .3 \text{ \AA}$

5.

For photons $\omega = ck \Rightarrow$ same dispersion
as plummens at low temperatures

✓ same as Debye with no upper limit !

$$U(T) = \int_0^\infty \frac{\Theta(\omega) \hbar\omega d\omega}{(e^{\hbar\omega/kT} - 1)} = \frac{V}{(2\pi^2 c^3)} \int_0^\infty \frac{\omega^2 \hbar\omega d\omega}{(e^{\hbar\omega/kT} - 1)}$$

$$= \frac{V}{(2\pi^2 c^3)} \frac{(kT)^4}{\hbar^3} \underbrace{\int_0^\infty \frac{x^3}{(e^x - 1)} dx}_{\pi^4/15}$$

Therefore

$$U(T) = \left(\frac{v}{2\pi^2} \right) \left(\frac{\pi^4}{15} \right) \frac{(kT)^4}{(hc)^3}$$



$$c_V = \frac{\partial U}{\partial T} = 4k_B \left(\frac{v}{2\pi^2} \right) \left(\frac{\pi^4}{15} \right) \frac{(kT)^3}{(hc)^3}$$



$$\boxed{\frac{c_V}{v} = \left(\frac{2\pi^2}{15} \right) \left(\frac{kT}{hc} \right)^3 k_B}$$

N.B.

$$\frac{c_V \text{ lattice}}{c_V \text{ photon}} \sim \left(\frac{c}{v} \right)^3 \sim \left(\frac{10^8 \text{ m/s}}{10^3 \text{ m/s}} \right)^3 \sim 10^{15}$$

where v is speed of sound

6.

$$\textcircled{a} \quad \omega = c_s k^2 \Rightarrow k = \left(\frac{\omega}{c_s} \right)^{1/2}$$

a) $D(\omega)$ Strategy : (i) Calculate $N(k)$ (2) Use dispersion relation $\Rightarrow N(\omega)$

$$(3) D(\omega) = \frac{dN(\omega)}{d\omega}$$

$$N(k) = \frac{\frac{4}{3} \pi k^3}{\frac{(2\pi)^3 / V}{6\pi^2}} = \frac{V}{6\pi^2} k^3$$

where $V = L^3$ 

$$N(\omega) = \frac{V}{6\pi^2} \left(\frac{\omega}{c_s} \right)^{3/2}$$



$$D(\omega) = \frac{dN(\omega)}{d\omega} = \frac{3}{2} \left(\frac{V}{6\pi^2 c_s} \right) \left(\frac{\omega}{c_s} \right)^{1/2}$$

$$D(\omega) = \left(\frac{V}{4\pi^2} \right) \frac{\omega^{1/2}}{\frac{c_s^{3/2}}{3}}$$

b)

$$N = \int_0^{\omega_{\max}} D(\omega) d\omega \quad \text{defines } \omega_{\max}$$



$$N = \left(\frac{V}{4\pi^2} \right) \frac{1}{c_s^{3/2}} \frac{\omega_{\max}^{3/2}}{3/2}$$



$$N = \left(\frac{V}{6\pi^2} \right) \frac{\omega_{\max}^{3/2}}{c_s^{2/2}}$$

$$(c) U(T) = \int_0^{\omega_{\max}} \frac{D(\omega) \hbar \omega d\omega}{\left(e^{\hbar \omega / kT} - 1 \right)}$$

$$= \left(\frac{V}{4\pi^2 c^{3/2}} \right) \int_0^{\omega_{\max}} \frac{\hbar \omega^{3/2} d\omega}{\left(e^{\hbar \omega / kT} - 1 \right)}$$

We note that

$$\frac{\partial}{\partial T} \left(e^{\frac{\hbar\omega}{kT}} - 1 \right)^{-1} = \frac{\hbar\omega}{kT^2} \frac{e^{\frac{\hbar\omega}{kT}}}{\left(e^{\frac{\hbar\omega}{kT}} - 1 \right)^2}$$



$$C_V = \frac{dU}{dT} = \frac{V}{4\pi^2 c^{3/2}} \frac{k^2}{kT^2} \int_0^{w_{\max}} \frac{w^{5/2} e^{\frac{\hbar\omega}{kT}}}{\left(e^{\frac{\hbar\omega}{kT}} - 1 \right)^2} dw$$



$$C_V = A \left(\frac{1}{kT^2} \right) (kT)^{5/2} \int_0^{\infty} \frac{x^{5/2} e^{-x}}{(e^{-x} - 1)^2} dx$$

B.

$$x = \frac{\hbar\omega}{kT}$$

$$C_V = \tilde{A} T^{3/2} \Rightarrow C_V \propto T^{3/2}$$

constant

7.

- Einstein considered a solid to be an ensemble of simple harmonic oscillators (SHOs) whose vibrational energies are quantized
- He assumed that they were independent with one frequency.
- At high temperatures he was able to recover the expected classical value of the specific heat (c_v).
- At low temperatures the Einstein specific heat was well below the classical value and
$$\lim_{T \rightarrow 0} c_v \rightarrow 0$$
- However Einstein's specific heat went to zero at low temperatures faster than what was observed experimentally.