The Universe: What We Know and What we Don't

Fundamental Physics

Cosmology
Elementary Particle Physics

Elementary Particle Physics Study of the small scale

- structure of the universe
- •What are the basic building blocks?
- How do they interact with one another?
- Is there a smallest amount of space and time?
- Is there a theory of everything?

Cosmology Study of the large scale structure of the universe

- •How big is the universe?
- •Where did the universe come from?
- •What is the fate of the universe?
- Are there other universes? How many

Particle-Wave Duality

Energy inversely proportional to wavelength





lower energy

Particle-Wave Duality







lower energy



higher energy

Large Hadron Collider



Study of small distances requires high energy probes

Large Hadron Collider



Energy scale 10^3 GeV Distance scale 10^{-19} m Temperature

 $10^{16} \mathrm{K}$

Big Bang

14 billion years ago the universe was much denser and hotter than today

Has been expanding and cooling ever since

To know the state of the universe at earlier and earlier times, need to know physics at higher and higher energy scales (smaller and smaller distances)

 $10^{16} \text{ K} \longrightarrow 10^{-12} \text{ s}$ after Big Bang

What we Know

• Physics down to a distance scale of

 10^{-19} m

• Physics down to a time of

$$10^{-12}$$
 s after the Big Bang

Fundamental Particle Physics

* What are the fundamental constituents of the universe?

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How do they interact with each other?

How to Judge How We're Doing

<u>Constituents</u>

- Number: economical
- Properties: few and simple
- Point-like? (no structure)

Theory

- Mathematically consistent
- Explains all observations
- Able to make predictions

Ancient Greece



Fundamental Physics



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Newton



Newton



Chemists Discover Evidence for Atoms

 NO_2

CH4

 $N_2 C$

1802



John Dalton

- Gay-Lussac's Law
- Boyle's Law
- Charles's Law
- Law of Multiple Proportions



World's First Particle Physicist

1827

Discovered Brownian Motion



Robert Brown

Botanist



Movie of Brownian Motion

Periodic Table

1869



Mendeleev

a classification scheme



Periodic Table

1869



Mendeleev

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Fundamental Particle Physics

End of 19th century

92 Atoms



Discovery of the Electron



J. J. Thomson



Thomson's CRT

A new particle electrically charged



Lord Rutherford

World's first high energy physicist

1910



Ernest Rutherford

very light electrons should have no effect on the alpha's

scattering of the alpha's will indicate structure of the "pudding"

Use high energy (5 MeV) alpha particles from radium decay to study structure of the atom.



The Nuclear Atom



Nearly all of the mass of the atom concentrated in a very small positively charged nucleus.

How small is the nucleus?

Discovery of the Neutron

1932



James Chadwick

Alpha particles interacting in air found to knock out neutral particles.

Rutherford had earlier discovered the proton (the nucleus of the hydrogen atom)



Atoms made out of: protons, neutrons, electrons

The Neutrino

- A free neutron decays to a proton and electron in about 15 minutes
 - not a 2-body decay
 - must be a third unseen particle

$$n \to p + e^- + \bar{\nu}$$

Ghost-like neutrino

Predicted in 1930 by Pauli Discovered in 1956 by Cowan and Reines



Momentum (MeV/c)



Structure of the Nucleus



Fundamental Particle Physics



Structure of the Proton



Mark 3 electron linac at Stanford University

The proton has a size it is not a point-like object

The Bevatron

6 GeV proton synchrotron in the hills of Berkeley









Designed to discover the anti-proton



Quarks

1964



Murray Gell-Mann



Stanford Linear Accelerator Center

SLAC 30 GeV electrons





2-mile long linear accelerator







Inside the Proton

SLAC









Antimatter

Carl Anderson discovers 1932 anti-electrons (positrons) positron track © Copyright California Institute of Technology. All rights reserved. Commercial use or modification of this material is prohibited.

Standard Model Particles



The Large Hadron Collider

