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HW# 2 solutions

Physics 406

① 8. Ch. 2, Q. 1

5/5 The scattered rays are nearly parallel because the detector is far away from the crystal, so the plane-wave approximation is valid.

② 8. Ch. 2, Q. 5

5/5 The lattice structure factors will be the same since they are both fcc. However, the diffraction patterns will have different spacings reflecting different unit cell dimensions.

③ 8. Ch. 2, Q. 7

5/5 No, there is no one-to-one correspondence. Reciprocal lattice vectors are normal to a set of crystal planes, which are defined by two direct lattice vectors.

④ 8. Ch. 2, Q. 9

5/5 For crystal diffraction we need $\lambda = \theta(1) \text{ \AA} \sim \text{interatomic spacing}$
Let's define $\lambda_0 = 1 \text{ \AA}$.

Using de Broglie relations,

$$\lambda_0 = \frac{h}{p} = \frac{h}{\sqrt{2mE}} \Rightarrow \text{Need } E = \frac{h^2}{2m\lambda_0^2}$$

Since $m_n > m_e$ $\left\{ \begin{array}{l} \uparrow \text{neutron} \\ \downarrow \text{electron} \end{array} \right.$ & $E \sim \frac{1}{m}$, $E_n < E_e$.

⑤ Q. Ch. 2, Pr. 1

Recall that

$$\lambda_{\min} = \frac{12.3}{V(\text{kV})} \text{ \AA} :$$

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$$V(\text{kV}) = \frac{12.3}{1.23} = \underbrace{10^4 \text{ V}}_{10 \text{ kV}}$$

The kinetic energy

$$KE = eV = \underline{\underline{10^4 \text{ eV}}}$$

6. θ Ch. 2, Pr. 2

Using $d_{hke} = \frac{2\pi}{G_{hke}}$ ^{proven in another problem} & the fact that the reciprocal of a cubic lattice is a cubic lattice, we obtain: $G_{hke} = \frac{2\pi}{a} \sqrt{h^2 + k^2 + l^2}$,

which gives

$$d_{hke} = \frac{a}{\sqrt{h^2 + k^2 + l^2}}$$

with $a = 2.62 \text{ \AA}$,

$$d_{100} = a = 2.62 \text{ \AA}, \quad d_{110} = \frac{a}{\sqrt{2}} = 1.85 \text{ \AA},$$

$$d_{111} = \frac{a}{\sqrt{3}} = 1.51 \text{ \AA}, \quad d_{200} = \frac{a}{2} = 1.31 \text{ \AA},$$

$$5/5 \quad d_{210} = \frac{a}{\sqrt{5}} = 1.17 \text{ \AA}, \quad d_{211} = \frac{a}{\sqrt{6}} = 1.07 \text{ \AA}.$$


Then the 1st order Bragg reflection gives $\sin \theta_{hke} = \frac{\lambda}{2d_{hke}}$.

with $\lambda = 1.54 \text{ \AA}$,

$$\sin \theta_{100} = 0.294 \Rightarrow \theta_{100} = 17.1^\circ$$

$$\text{Likewise, } \begin{cases} \theta_{110} = 24.6^\circ, & \theta_{210} = 41.1^\circ, \\ \theta_{111} = 30.6^\circ, & \theta_{211} = 46.0^\circ, \\ \theta_{200} = 36.0^\circ, \end{cases}$$

Note that smaller interplanar distances correspond to larger angles.

7.  0. Ch. 2, Pr. 3

a) $\lambda = 1.54 \text{ \AA}$

$\theta = 19.2^\circ$

(111) planes

Bragg's law: $n\lambda = 2d \sin \theta$.

5/5 assuming $n=1$,

$$d_{111} = \frac{\lambda}{2 \sin \theta} = \frac{1.54 \text{ \AA}}{2 \sin(19.2^\circ)} = \underline{\underline{2.34 \text{ \AA}}}$$

b)

$$\rho = \frac{\# \text{ Al atoms}}{\text{unit cell volume}} \times \frac{\text{mol. weight of Al}}{N(\# \text{ atoms in 1 mole})}$$

Al is fcc: $8 \times \frac{1}{8} + 6 \times \frac{1}{2} = 4 \frac{\text{atoms}}{\text{unit cell}}$.

$V = a^3$ unit cell volume

⑧ The plane (hkl) is defined by its intercepts

$$5/5 \quad \frac{\vec{a}_1}{h}, \quad \frac{\vec{a}_2}{k}, \quad \frac{\vec{a}_3}{l}$$

(a) Define
$$\begin{cases} \vec{A} = \frac{\vec{a}_1}{h} - \frac{\vec{a}_2}{k}, \\ \vec{B} = \frac{\vec{a}_1}{h} - \frac{\vec{a}_3}{l} \end{cases}$$

These 2 vectors are in the plane.

$$\vec{G} = h\vec{a}_1 + k\vec{a}_2 + l\vec{a}_3 =$$

$$= \frac{2\pi}{V} h (\vec{a}_2 \times \vec{a}_3) + \frac{2\pi}{V} k (\vec{a}_3 \times \vec{a}_1) +$$

$$+ \frac{2\pi}{V} l (\vec{a}_1 \times \vec{a}_2), \text{ where } V \text{ is the unit cell volume.}$$

Then
$$\vec{G} \cdot \vec{A} = 2\pi - 2\pi = 0,$$

$$\vec{G} \cdot \vec{B} = 0.$$

\vec{G} is \perp to (hkl) .

(b) Let \hat{n} be the unit normal to the plane (hkl) .

$$\frac{\vec{a}_1}{h} \cdot \hat{n} = d_{hkl} \quad (\text{projection of } \frac{\vec{a}_1}{h} \text{ onto } \hat{n})$$

But $\hat{n} = \frac{\vec{G}}{|\vec{G}|}$, as shown in (a).

$$\text{Then } d_{hke} = \frac{\vec{a}_1 \cdot \vec{G}}{h|\vec{G}|} = \frac{2\pi}{|\vec{G}|} =$$

(c) For an sc lattice,

$$\vec{G} = \frac{2\pi}{a} (h\hat{x} + k\hat{y} + l\hat{z})$$

$$d_{hke}^2 = \frac{4\pi^2}{G^2} = \frac{4\pi^2}{\left(\frac{2\pi}{a}\right)^2 (h^2 + k^2 + l^2)} =$$
$$= \frac{a^2}{h^2 + k^2 + l^2} =$$



⑨ Key points of Handout 3:

(a) Oblate spheroids pack more densely than do spheres when poured randomly & shaken.

(b) More experiments with different spheroid aspect ratios & computer simulations show that random

5/5 packing densities ^{can} approach the perfect packing ratio of 0.74.

(c) It was always assumed that periodic orderings are denser than random ones \Rightarrow true for spheres but may not be true for all spheroids.

(d) Changes ~~in~~ in spheroid shape may lead to major changes in random packing densities.