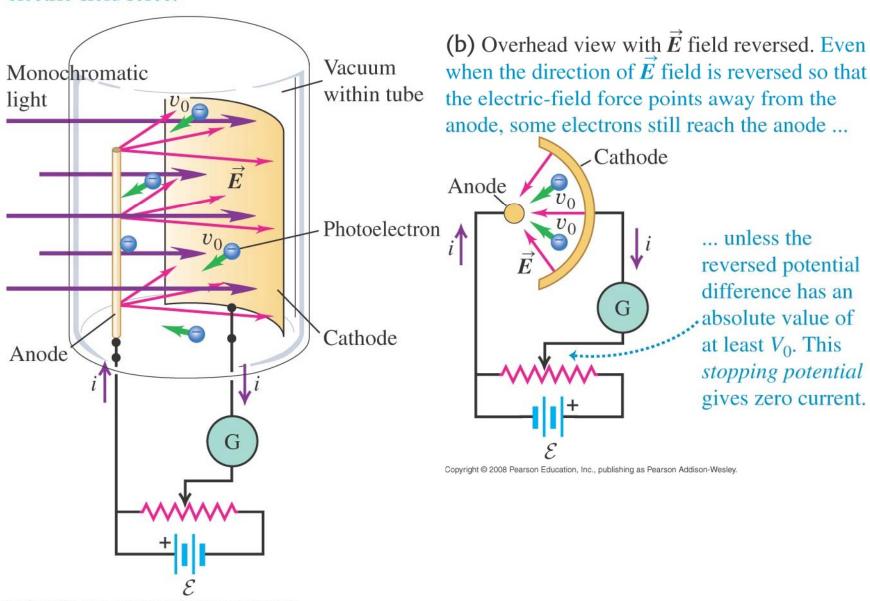
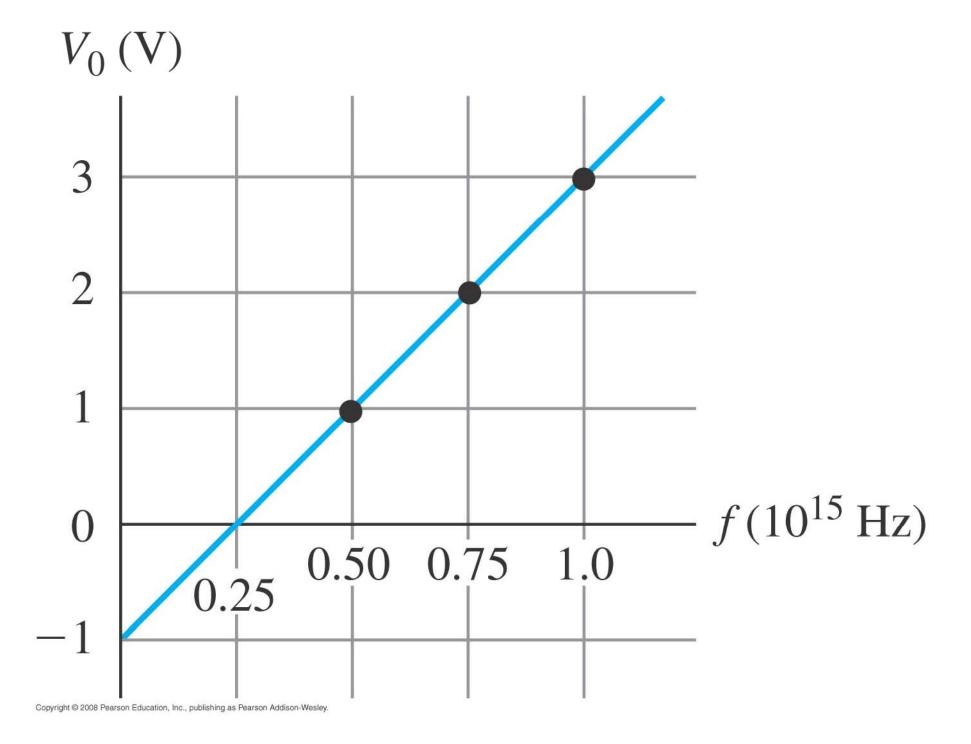
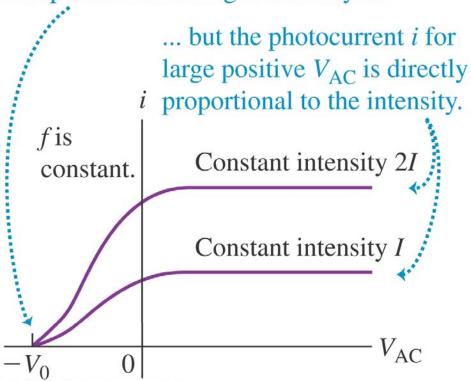
(a) Light causes the cathode to emit electrons, which are pushed toward the anode by the electric-field force.

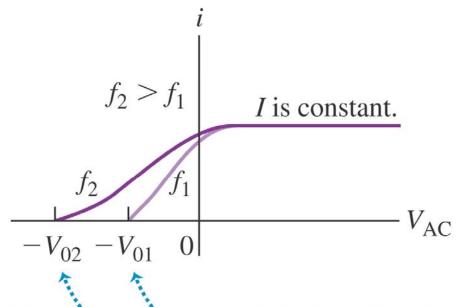


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The stopping potential V_0 is independent of the light intensity ...





The stopping potential V_0 (and therefore the maximum kinetic energy of the photoelectrons) increases linearly with frequency: since $f_2 > f_1$, $V_{02} > V_{01}$.



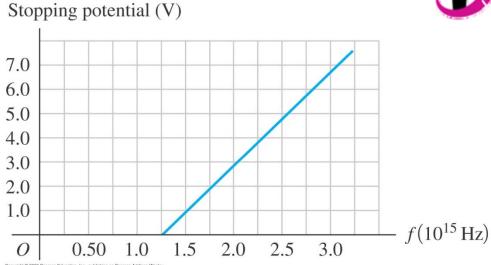
In an experiment to demonstrate the photoelectric effect, you shine a beam of monochromatic blue light on a metal plate. As a result, electrons are emitted by the plate.

If you increase the intensity of the light but keep the color of the light the same, what happens?

- A. More electrons are emitted per second.
- B. The maximum kinetic energy of the emitted electrons increases.
- C. both A. and B.
- D. neither A. nor B.

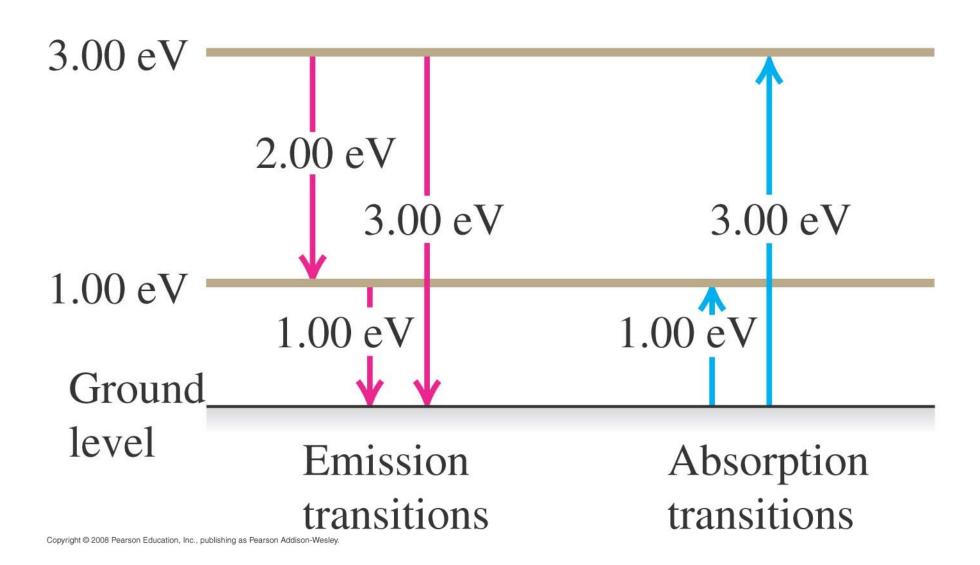
0

This graph in shows the stopping potential as a function of the frequency of light falling on a metal surface. If a different type of metal is used,



- A. the graph could have a different slope.
- B. the graph could intercept the horizontal axis at a different value.
- C. both A. and B.
- D. neither A. nor B.







A certain atom has two energy levels whose energies differ by 2.5 eV.

In order for a photon to excite an electron from the lower energy level to the upper energy level, what must be true about the energy of the photon?

- A. Its energy must be greater than or equal to 2.5 eV.
- B. Its energy must be exactly 2.5 eV.
- C. Its energy must be less than or equal to 2.5 eV.
- D. none of the above